

1 **Identification of contamination sources through the**
2 **application of nitrogen, oxygen and deuterium isotopes in**
3 **the Estarreja shallow aquifer, Aveiro (Portugal)**

4 Ana Carolina Pires Marques¹, Maria do Rosário Carvalho², Eduardo Anselmo Ferreira
5 da Silva¹

6 ¹ GeoBioTec Research Centre, Department of Geosciences, University of Aveiro, Campus
7 Universitário de Santiago, 3810-193, Aveiro, Portugal

8 ² Department of Geology, Faculty of Sciences & IDL University of Lisbon, Campo Grande,
9 1749-016, Lisboa, Portugal

10 a.carolina.marques@hotmail.com

11 **Abstract.** Agriculture, chemical industry and livestock activity are the most
12 important activities in the Estarreja municipality and have put at risk the quality
13 of surface water and groundwater, leading to a great vulnerability of the Estar-
14 reja shallow aquifer. In this respect, the main goal of the present study is the
15 identification of contamination sources through the application of nitrogen and
16 oxygen isotopes, beside the deuterium and oxygen isotopes in the Estarreja
17 shallow aquifer. Results of $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ in the nitrate indicate isotopic signa-
18 tures for ammoniacal fertilisers, wastewater and animal manure, clearly show-
19 ing the influence of industrial and agricultural activities. The $\delta^2\text{H}$ and $\delta^{18}\text{O}$ in
20 the water enrichment of some samples could suggest salinization phenomenon.

21 **Keywords:** Estarreja; Groundwater; Nitrate; N origin; Stable isotopes

22 **1 Introduction**

23 The Nitrate Directive (91/676/EEC) concerns about the protection of the water quality
24 against pollution caused by nitrate from agricultural sources. It aims to reduce the
25 quantities of nitrate in water and prevent its continued rise. Additionally, the Urban
26 Wastewater Directive (91/271/EEC) aims to protect the environment from the harm-
27 ful effects caused by wastewater discharges. The implementation of Nitrate Directive
28 and Water Framework Directive (2000/60/EC) led to the designation of large areas
29 vulnerable to nitrate pollution, designated as Nitrate Vulnerable Zones (NVZs). NVZs
30 are areas at risk from agricultural nitrate pollution in which groundwater contain or
31 could contain (if no action is taken to reverse trend) more than 50 mg/L of NO_3 . The
32 objectives are to ensure that the nitrate concentration in surface water and groundwa-
33 ter does not exceed the limit of 50 mg/L, as well as to control the incidence of the
34 eutrophication phenomenon. The NVZ of Estarreja-Murtosa, Aveiro is the study case
35 area to consider. In Estarreja region, the groundwater shows nitrate concentration of
36 up to 140 mg/L. Agriculture, chemical industry and livestock activity are the most

37 important activities in the Estarreja municipality with a perfect symbiosis between
 38 them since the 1930s. Around the 1950s, the installation of the Estarreja Chemical
 39 Complex (ECC) took place with a huge investment mainly in the production of am-
 40 monia for the manufacture of nitrogen fertilisers used in agriculture.

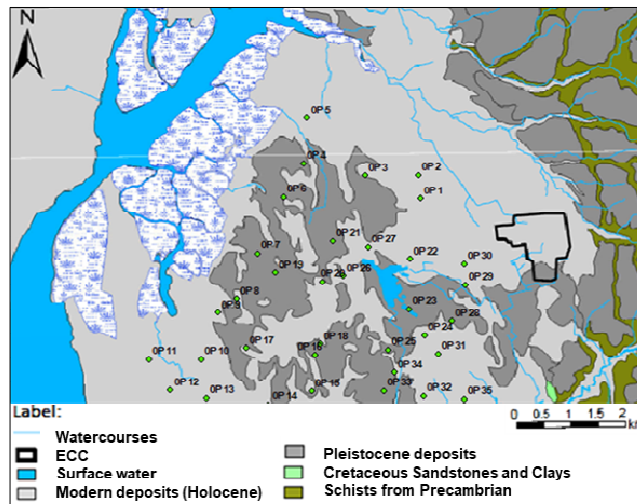
41 The literature has described some approaches in order to distinguish the different
 42 sources of nitrate in groundwater, generally with $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ in the NO_3 . The $\delta^2\text{H}$
 43 and $\delta^{18}\text{O}$ in the H_2O are useful to understand the hydrological cycle, especially the
 44 aquifer recharge, as well as salinization phenomenon. The present study aimed to
 45 discriminate the contamination sources through the application of these two isotopic
 46 couples in water samples from the Estarreja shallow aquifer.

47 2 Materials and Methods

48 The study area is characterized by Holocene and Pleistocene detrital sedimentary
 49 deposits [1]. This area has two important aquifer systems [2]: (a) the Aveiro Creta-
 50 ceous Multi-Aquifer System (ACMAS) and (b) the Aveiro Quaternary Aquifer Sys-
 51 tem (AQAS) which is essentially made up of three main aquifer units with different
 52 hydrogeological and hydraulic characteristics between them [3]. The upper aquifer is
 53 laid in modern deposits from Holocene that comprise dune formations or dune sands
 54 and alluvial deposits [4]. The second aquifer unit, a semi-confined aquifer, is laid in
 55 Pleistocene deposits, at Quaternary base aquifer [3]. The third aquifer unit is made up
 56 of Cretaceous Sandstones and Clays and Schists from Precambrian (before 540 Ma).
 57 One field campaign was carried out in Estarreja for water sampling at wet season (4th
 58 and 5th May 2018). A total of 35 wells or small diameter holes were sampled with a
 59 maximum depth of 7 meters at the Estarreja shallow aquifer (Figure 1). The $\delta^2\text{H}$ and
 60 $\delta^{18}\text{O}$ in the H_2O measurements (vs. V-SMOW – Vienna-Standard Mean Ocean Wa-
 61 ter) were performed by Mass Spectrometry at the Stable Isotopes and Instrumental
 62 Analysis Facility (SIIAF) in the Faculty of Sciences, University of Lisbon (Portugal).

63 The nitrate extraction
 64 as well as $\delta^{15}\text{N}$ and
 65 $\delta^{18}\text{O}$ in the NO_3 deter-
 66 mination were carried
 67 out in frozen samples
 68 shipped to the Envi-
 69 ronmental Isotope
 70 Laboratory (EIL) in the
 71 University of Waterloo
 72 (Canada), and analysed
 73 with a continuous flow
 74 isotope-ratio mass
 75 spectrometer, with
 76 precision of 0.5‰.

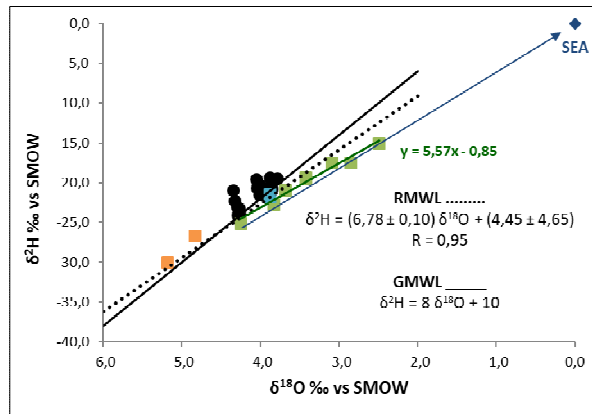
77 **Fig. 1.** Geologic
 78 Framework [3]



79 3 Results and Discussion

80 3.1 $\delta^2\text{H}$ and $\delta^{18}\text{O}$ in the H_2O

81 The sampled groundwater has meteoric origin considering the Global Meteoric Water
 82 Line (GMWL, [5]) represented by the equation $\delta^2\text{H} = 8 \delta^{18}\text{O} + 10$ or the Regional
 83 Meteoric Water Line in Portugal (RMWL, [6]) with isotopes relation following the
 84 equation $\delta^2\text{H} = (6.78 \pm 0.10) \delta^{18}\text{O} + (4.45 \pm 4.65)$, $R = 0.95$, because all the samples
 85 are plotted around the lines [6] and all the deviations can be explained by physical
 86 processes as the slope of the evaporation line is 5.58, a value between 3 and 6. The
 87 samples show excess of deuterium (d) values are between 4.9 ‰ and 13.8 ‰. In Fig-
 88 ure 2, it is possible to identify different groups of water: $\delta^2\text{H}$ isotopes enriched sam-
 89 ples probably associated to a great humidity at the recharge area (black dots); samples
 90 enriched in heavy isotopes showing a shift to the right of the water lines that could be
 91 affected by evaporation or mixed with seawater (blue and green dots); samples de-



pleted in heavy isotopes can be related with far way recharge area, representing condensation phenomena (orange dots).

Fig. 2. Isotopic composition of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ from the Estarreja shallow aquifer and its location in relation to the GMWL [5] and the RMWL [6].

104 3.2 $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ in the NO_3

105 The dissolved nitrate in groundwater have three different types of $\delta^{15}\text{N}$ isotopic signa-
 106 ture: overlap between wastewater and animal manure (Figure 3, $10 \leq \delta^{15}\text{N} \leq 20$ [7]
 107 [8]), overlap between ammoniacal fertilisers and wastewater ($5 \leq \delta^{15}\text{N} \leq 8$, [8]) and a
 108 typical signature of wastewater ($8 \leq \delta^{15}\text{N} \leq 10$, [8] [9]).

109 4 Conclusions

110 In terms of $\delta^2\text{H}$ and $\delta^{18}\text{O}$ in the H_2O , the samples represented by blue and green dots
 111 samples can suggest a salinization phenomenon as they can be mixed with seawater,
 112 assuming $\delta^2\text{H}$ and $\delta^{18}\text{O}$ values for seawater are close to zero. In relation to $\delta^{15}\text{N}$ and
 113 $\delta^{18}\text{O}$ in the NO_3 , it is possible to conclude that the samples with an isotopic signature
 114 of overlap between wastewater and animal manure could indicate clearly an agricul-

115 ture component, in contrast to the samples with an isotopic signature of overlap be-
 116 tween ammoniacal fertilisers and wastewater which can suggest an industrial and/or
 117 an agriculture component. The isotopic signature of ammoniacal fertilisers can have
 118 its origin as industrial products or as fertilisers applied in soils.

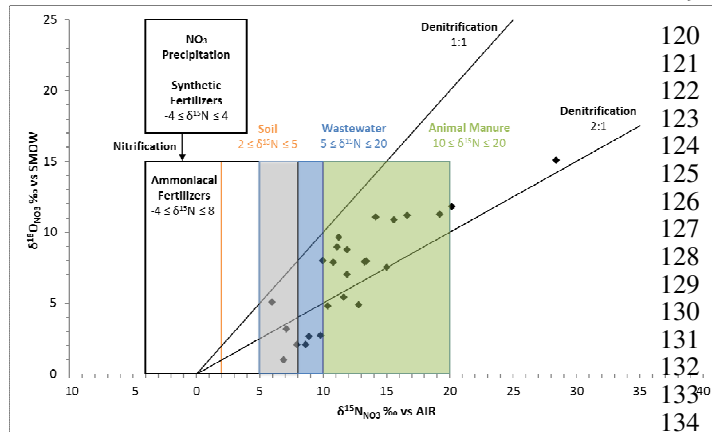


Fig. 3. Stable isotope composition of dissolved nitrate in groundwater: $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$; fields are according to bibliography (see 3.2 Results and Discussion); the trends resulting from nitrate transformation processes, nitrification and denitrification

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